BBA 42968

Review

Molecular aspects of the energetics of nitrogen fixation in *Rhizobium*-legume symbioses *

Mark R. O'Brian ** and Robert J. Maier

Department of Biology, The Johns Hopkins University, Baltimore, MD (U.S.A.)

(Received 17 June 1988)

Key words: Nitrogen fixation; ATP synthesis; Hydrogen oxidation; Electron transport; Leghemoglobin

Contents	
I. Introduction	230
II. Carbon metabolism	230
A. Physiological studies	230
B. Analyses of mutants	230
C. Genetics	231
D. Carbon metabolism in free-living Rhizobium	232
III. Leghemoglobin	232
A. Properties and function	232
B. The origin of leghemoglobin heme	233
C. Regulation of heme synthesis	235
D. Heme as a regulator	235
IV. Electron transport	236
A. Cultured cells	236
B. Bacteroids	236
V. ATP synthesis and ion transport	237
A. ATP synthesis in N ₂ -fixing cells	238
B. Ion transport	239
1. Potassium	239
2. Molybdenum	239
3. Nickel	239
4. Ammonium	240
VI. Hydrogen oxidation	240
A. Regulation	240
B. Components of H ₂ -oxidation	241
C. Genetics	242
Note added in proof	242
Acknowledgements	242
References	242

Abbreviations: ALA, δ-aminolevulinic acid; Hup, hydrogen uptake.

Correspondence: R.J. Maier, Department of Biology, The Johns Hopkins University, Baltimore, MD 21218, U.S.A.

^{*} The term 'Rhizobium' without a species name refers collectively to the genera Rhizobium, Bradyrhizobium and, with some reluctance, Azorhizobium, all of which form symbioses with certain legumes.

^{**} Present address: Department of Biochemistry S.U.N.Y. at Buffalo, Buffalo, NY 14214, U.S.A.

I. Introduction

Biological nitrogen fixation is an energy-intensive process; the amount of ATP used per N_2 fixed can be as high as 42 according to some growth yield measurements [203]. Under ideal conditions N_2 fixation via nitrogenase proceeds according to the following reaction:

$$N_2 + 8H^+ + 8e^- + 16ATP \rightarrow 2NH_3 + H_2 + 16(ADP + P_i)$$
 (1)

The discrepancy in the ATP expenditure between the measured values and Eqn. 1 is that the latter does not take into account the cost of the high-energy electrons, nor does it consider that the ATP-dependent H₂ evolution can exceed N₂ reduction stoichiometrically under experimental and physiological conditions. Sixteen ATP molecules are hydrolyzed per eight electrons transported through the nitrogenase complex, and thus H⁺ reduction diminishes the efficiency of N₂ fixation. Chemical N₂ reduction to ammonia requires energy in the form of heat and pressure, but it also requires the exclusion of O₂ and a nonaqueous environment [174], and hence it is miraculous that biological N₂ fixation occurs at all. This communication will not cover the enzymology of the nitrogenase complex, nor will it address the ATP requirements for nitrogenase at the level of that enzyme (these topics are reviewed in Refs. 166, 174 and 197). Rather, we wish to begin the discussion with the understanding that N₂-fixed bacteria need to synthesize a great deal of ATP and reducing equivalents, and proceed directly to problems in energetics which are unique to symbiotic N₂-fixing organisms.

The N₂-fixing root nodule in legumes is formed from the symbiosis of a Rhizobium or Bradyrhizobium species with a particular plant host. Nitrogen fixation is carried out by the bacteroids *, and they are also the direct source of ATP which drives the reaction. Although plant mitochondria in nodules almost certainly cannot synthesize the ATP necessary for N_2 fixation [175], the plant host is involved in every aspect of the energetics of this process. The carbon sources oxidized by the bacteroids for ATP synthesis are ultimately products of photosynthesis, and the leghemoglobin found in nodules, which prevents O₂ inactivation of nitrogenase, is a plant product. Also, the host plant apparently has some control over the efficiency of N₂ fixation at the enzyme level [22-24], and it can affect the expression of the rhizobial hydrogenase [22-24,40,111,151]. It is obvious then that strategies adopted to accommodate the restrictions and peculiarities of N₂ fixation are different in symbiotic associations than in free-living organisms.

II. Carbon metabolism

II-A. Physiological studies

Unlike cell growth in culture, the carbon sources available to *Rhizobium* cells in root nodules are not subject to the whim of the investigator, and thus knowledge of the carbon sources which fuel nitrogen fixation symbiotically has not been easily ascertained. The energy for nitrogen fixation is ultimately derived from plant host photosynthate, and high concentrations of carbohydrate are found in nodules, but these compounds are not the immediate carbon-sources driving nitrogen fixation.

Carbon compounds have been tested for their ability to be taken up by isolated bacteroids [79,95,176,188, 189,193] or for their ability to support respiration and nitrogen fixation [27,171,216,217]. An early study by Tuzimura and Meguro [219] noted that Bradyrhizobium japonicum bacteroid respiration was enhanced by the addition of succinate to a much greater extent than by the addition of glucose, and thus they speculated that organic acids rather than carbohydrates may be the energy sources in symbiotically grown cells. Direct measurements of nitrogenase activity support this view [27,171,217], and although exceptions have been noted [216,217], it is generally believed that carbohydrate oxidation is not a major energy-source driving nitrogen fixation [27,188]. Carbohydrate uptake is inducible in free-living cells, but the symbiotic influx rate is slow compared with that of succinate or malate [95,176,188], and it is a passive process, at least in B. japonicum [176]. The uptake of C₄-dicarboxylic acids by bacteroids has been readily demonstrated [79,95,176,188,189,192], and the influx is an active process [176,189,192]. The transport of succinate, fumarate and malate shares a common system in bacteroids and in cultured cells of Rhizobium leguminosarum [72,79], Rhizobium meliloti [67], and B. japonicum [138,189], as discerned from kinetic data, and also from the study of mutants (subsection IIB).

In addition to dicarboxylic acids, there is evidence that glutamate is also involved in bacteroid respiration in *B. japonicum* [187]. Although succinate and malate are transported into *B. japonicum* more rapidly than is glutamate, the proportion of glutamate converted into CO₂ is substantial compared to the dicarboxylic acids. In fact, much of the succinate and malate (as well as aspartate) transported into cells is converted into glutamate.

IIB. Analyses of mutants

The generation of mutants in carbon uptake and metabolism have confirmed many of the physiological studies, and they have provided considerable insight

Bacteroids is the term given to the differentiated, N₂-fixing bacteria found in root nodules.

into the carbon requirements of symbiotically grown cells. Ronson and Primrose [186] generated carbohydrate metabolism mutants of Rhizobium trifolii with lesions in glucokinase, fructose uptake, the Entner-Doudoroff pathway and in pyruvate carboxylase. These mutant strains formed effective symbioses with red clover, suggesting that neither glucose, nor fructose nor sucrose is required for nitrogen fixation by bacteroids, and thus the ATP and reductant required for this process must come from other carbon sources. Subsequent mutant analyses showed that C4-dicarboxylic acids are required for effective clover nodule formation. Rhizobium trifolii mutant strains unable to grow on, or transport, succinate, fumarate or malate still nodulated clover, but those nodules were ineffective [183]. The isolation of bacterial mutants from R. meliloti [36,67,71,74] and R. leguminosarum [16,73,76] also demonstrate the importance of C₄-dicarboxylic acids in symbiotic nitrogen fixation. It should not be overlooked that these dicarboxylic transport (dct) mutants do form nodules that are invaded with bacteria, and therefore organic acids are not essential for bacterial growth symbiotically. It is not known what carbon sources are required for cell growth and maintenance as R. leguminosarum mutants defective in sugar metabolism [54,78] or strains with double mutations in organic acids and carbohydrate metabolism [15] still form root nodules on peas, and lesions in gluconeogenesis form effective nodules as well [140]. It seems that other carbon sources are available to the nodule, or that the lesions in the carbon mutants are circumvented by alternate pathways or by the ability of the plant host to rescue the mutant. Saroso et al. [191] argued that the peribacteroid membrane may be impermeable to sugars, since snake bean nodule bacteroids do not express enzymes which are inducible by sucrose in culture despite the presence of high concentrations of sucrose in nodules.

The transport of the dicarboxylic acids succinate, malate and fumarate occur via a common system as deduced from mutant analyses [36,67,183,226], which agrees with the kinetic data described above (subsection IIA). Mutants isolated as succinate transport mutants are invariably deficient in the uptake of other dicarboxylic acids, but they are able to incorporate carbohydrates as well as do the respective wild-type strains. These mutants, isolated in culture, are also *Dct* deficient as bacteroids, and thus it seems that C₄-dicarboxylic acids are transported by the same system under both growth conditions. Exceptions to this are noted below (subsection IIC) whereby Tn 5-induced mutants in regulatory *dct* genes show different phenotypes symbiotically compared with cultured cells.

IIC. Genetics

Ronson et al. [181] isolated cosmids containing C₄-dicarboxylic acid transport (dct) genes by complement-

ing putative structural and regulatory dct mutants of R. trifolii with an R. leguminosarum gene bank. Three contiguous dct gene loci were identified within a 5.5 kb region of DNA, as discerned from mapping and transposon mutagenesis of subcloned fragments of the isolated cosmids. The dctA gene locus not only complemented a putative structural dct mutant of R. trifolii when expressed from its own promoter, but it also complemented a regulatory dct mutant of R. trifolii and of E. coli (dctB) when expressed from a constitutive vector promoter; these observations strongly suggest that the dctA locus codes for a structural gene. The dctA gene codes for a hydrophobic protein, and it is apparently the only structural gene necessary for C₄-dicarboxylic acid transport in Rhizobium grown symbiotically and in culture [180].

dctB and dctD code for positive regulatory elements which are necessary for the expression of the dctA gene product, and they are transcribed divergently from dctA [181]. Constitutive expression of dctD from a vector promoter complemented mutants with lesions in dctB which suggests that DctD expression may normally require the dctB gene product. The DNA sequences of the dct genes [180,182], and the inferred amino acid sequences, have revealed very interesting clues as to how C₄-dicarboxylic acid transport is regulated in Rhizobium. The dctA promoter is homologous to the consensus promoter sequence -26 CTGGYAYR-N_a--TTGCA-10 found in genes which are under control of the ntrA (also called rpoN) gene product [180]. The ntrA, as well as ntrB and ntrC, gene product regulates the expression of genes involved in the assimilation of nitrogen under nitrogen-limiting conditions in enteric bacteria [118,130]. Ronson et al. [184] isolated an ntrAlike gene from R. meliloti using a scheme which assumed that the ntrA gene product (NtrA) does indeed regulate C₄-dicarboxylic acid transport. The deduced amino acid sequence of the R. meliloti NtrA shares 38% homology with NtrA from Klebsiella pneumoniae, and it is required for nitrate assimilation and nitrogen fixation, as well as for C₄-dicarboxylic acid transport [184].

The DNA sequences of dctB and dctD [182] show that DctB and DctD are homologous to NtrB and NtrC, respectively, and are in fact part of a conserved family of two-component regulatory systems which respond to stimuli from the environment [155,185,229]. It is thought that one component of these systems (constitutively expressed at a low level) acts as a sensor, and the second component receives a signal from the sensor which then activates the response (Fig. 1). According to this model, the sensor protein DctB is activated upon binding to a C₄-dicarboxylic acid which in turn activates DctD. Activated DctD would then act in concert with NtrA in the transcription of dctA. By analogy to the Ntr systèm [154], DctB may phosphorylate DctD in order to activate it; this idea is supported by the finding that

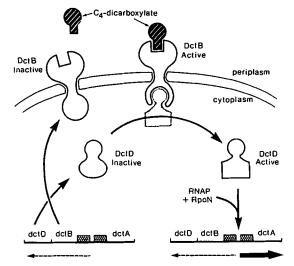


Fig. 1. Model for the activation of the dctA gene. Reprinted with permission from Ronson et al. (1987) Nucleic Acids Research 15, 7921.

DctD has a potential ATP-binding site in a region that is strongly conserved in NtrC [182]. However, the constitutive transcription of the cloned dctD gene by a vector promoter can overcome the need for the dctB gene product in C_4 -dicarboxylic acid transport [181], thus DctD can be activated by a factor other than DctB, or else DctD can activate itself under those experimental conditions.

Most of the detailed genetic studies of C₄-dicarboxylic acid transport described above have been done with R. leguminosarum, but there are good reasons to believe that the Dct system is similar in other Rhizobium species. Regulatory, as well as structural, R. leguminosarum dct genes complement dct mutants of R. trifolii [181], and they can be regulated in R. meliloti [184]. dct genes isolated from R. meliloti [36] can be expressed in B. japonicum [33], and they confer increased ex planta nitrogen-fixing ability on the recipient strain. Watson et al. [226] identified two dct regions on a megaplasmid of R. meliloti. Interestingly, Tn 5-induced mutants in one locus yielded ineffective nodules whereas mutations in the second locus resulted in nodules with wild type levels of nitrogen fixation activity even though mutations in either locus were Dct in culture. These findings indicate that C₄-dicarboxylic acid transport may be regulated by different factors under free-living and symbiotic growth. Likewise, Ronson et al. found that Tn5-induced dctB and dctD mutants form nodules with good nitrogenase activity [182], although the chemically mutagenized dctB gene results in nodules which cannot fix nitrogen [183]. The latter group [182] speculated that the apolar mutation in the dctB gene caused by chemical mutagenesis would result in the synthesis of inactive DctD which could bind to the dctA promoter, and thus inhibit transcription by the symbiotic regulator. The polar mutation in

the dctB gene, however, would result in no synthesis of DctD, since dctD is downstream from dctB, and thus the symbiotic regulator could activate dctA transcription.

IID. Carbon metabolism in free-living Rhizobium

Carbon metabolism in Rhizobium cultured cells has been thoroughly reviewed [204], and only a brief mention of it is presented here. It is a topic in itself not only because carbon metabolism is quite different from that of symbiotically grown cells, but also because there are significant differences between the fast- and slow-growing species [204,235], and there are even differences among strains of the same species [119]. Since free-living species do not, for the most part, fix nitrogen, the rationale for understanding carbon metabolism in these cells is not so much from the perspective of its importance in supporting nitrogen fixation directly. Rather, it is useful in studying systems and pathways that are common to both growth modes, for understanding how cells change metabolically in the free-living and symbiotic states, and in understanding how Rhizobium behaves in the rhizosphere in general.

Fast-growing rhizobia can grow on hexoses, pentoses, disaccharides, trisaccharides and organic acids, whereas the slow-growing species do not grow on disaccharides or trisaccharides [204]. Growth on organic acids and sugar alcohols is variable among strains, and glycerol is the most universal carbon source. Fast- and slow-growing *Rhizobium* species are able to grow on a wide variety of aromatic compounds [167,204], which may be significant for survival in the soil.

III. Leghemoglobin

IIIA. Properties and function

Leghemoglobin is found in the root nodules of all legumes, and plant hemoglobins have also been identified in the non-legume symbioses of *Bradyrhizobium* sp. with Parasponia [12] and in the so-called 'actinorhizal' symbioses of Frankia with various plant hosts [212]. The leghemoglobin apoprotein is a plant product [20,198,225], whereas the heme prosthetic group may be synthesized by the bacterium (subsection IIIB). Leghemoglobin has been intensively studied genetically, physically and physiologically (reviewed in Refs. 7, 35 and 53), and it is a fascinating protein in many respects. Analysis of two soybean leghemoglobin genes show that each contains two introns in the same positions as those of animal hemoglobins [98], and the amino acid sequence of leghemoglobin also shows homology to animal globins in critical regions [96]. Since hemoglobin-like DNA sequences have also been found in non-nodulating plant species, Appleby et al. [10] speculated that hemoglobins may occur in all plants, and that their origin, along with animal hemoglobin genes [8], are descended from a common ancestor.

A problem inherent to aerobic nitrogen-fixing organisms is that O₂ is essential for ATP synthesis, but the nitrogenase enzyme is O₂-labile, and thus a strategy must be adopted to deal with this paradox. The leghemoglobin found in root nodules facilitates the diffusion of O₂ to the bacteroids and buffers the free O₂ concentration in the nodule at a very low tension (reviewed in Refs. 8 and 230). The O₂ can be poised at a low concentration in nodules because leghemoglobin has a very high affinity for O₂. The dissociation binding constant of soybean leghemoglobin is 46 nM [99,231], and if an average partial oxygenation of 20% is assumed, then the free O₂ concentration in soybean nodules is 11 nM [8]. This value is reasonably consistent with the apparent dissociation binding constant of 6 nM of the efficient bacteroid oxidase for O₂ [30], and thus ATP-producing respiration can occur at the very low O₂ tension in the nodule. Also, if the leghemoglobin concentration in soybean nodule cytosol is 3 mM [26], and the other conditions described above are true, then the ratio of bound to free O₂ is 55 000 [8]. This means that leghemoglobin provides an enormous O2 buffering capacity, and that the low free O₂ concentration can be maintained despite the rapid rate of bacteroid respiration. The high leghemoglobin concentration also allows the diffusion of O₂ to be facilitated by the much larger globin molecule. The high affinity of leghemoglobin for O_2 is due to a very fast association rate ($k_a = 1.2 \cdot 10^8$ M⁻¹·s⁻¹ for soybean leghemoglobin, Refs. 11, 99 and 231), rather than a very slow O₂ dissociation rate. The O_2 dissociation rate constant is 5.5 s⁻¹ [11], which is fast enough for soybean leghemoglobin to be kinetically competent to carry O_2 to the bacteroids [8].

Leghemoglobin was implicated in facilitated O_2 diffusion when Wittenberg et al. [232] found that the addition of leghemoglobin to isolated bacteroids resulted in a 10-fold increase in nitrogenase activity with only a 2-fold increase in the rate of O_2 uptake. Other O_2 -binding proteins could substitute for leghemoglobin provided that the O_2 association and dissociation rates were sufficiently fast to support respiration. They postulated that leghemoglobin facilitates O_2 diffusion through the 'unstirred layer' surrounding the suspended bacteroids, and buffers O_2 at a concentration favoring electron transport through the efficient limb of a branched electron transport system (subsection IVB).

The nitrogenase enzyme complex is very well conserved and the O_2 lability of the purified proteins probably do not vary much between organisms. However, aerobic N_2 -fixers show a range of sensitivities to O_2 , and this is also true among the rhizobia. Azorhizobium caulinodans has very high nitrogenase activity in cultured cells grown in 2 μ M O_2 [60,203], whereas B.

japonicum bacteroids have no nitrogenase activity at this O₂ tension. Lupin bacteroids have nitrogenase activity when isolated from nodules aerobically [136], but activity in soybean bacteroids is irreversibly abolished in air. R. leguminosarum bacteroids have an O₂ optimum of 800 nM for nitrogenase activity [120], but B. japonicum bacteroids are not very active at this O₂ concentration, and have an O2 optimum closer to 100 nM [28]. The low O_2 concentration optimum for B. japonicum nitrogenase activity is probably a direct result of ATP synthesis being more efficient at 100 nM O₂ than at 800 nM O₂ [29], which in turn is a function of the O₂ affinity of the terminal oxidases [30]. R. leguminosarum retains the 'free-living' oxidases cytochromes aa₃ and o symbiotically [9], which may reflect a higher free O₂ concentration in pea nodules compared with that of soybean nodules, where cytochromes aa₃ and o are repressed. B. japonicum has apparently evolved as an O_2 -intolerant species compared with R. leguminosarum and Bradyrhizobium sp. (Lupinus) as judged by the O₂ lability of nitrogenase and by the low O₂ concentration at which maximal ATP synthesis and nitrogen fixation occurs in B. japonicum. Appleby [7] noted that the O₂ tolerance of R. leguminosarum compared with that of B japonicum is also reflected in the binding affinity of the respective leghemoglobins for O_2 . The O₂ dissociation binding constant of soybean leghemoglobin is 46 nM [99,231] and 149 nM for pea leghemoglobin [221], and thus the pea hemoglobin would buffer O₂ at a higher concentration than is found in soybean nodules. Also, lupin bacteroid nitrogenase activity is greater at the myoglobin-buffered O₂ concentration than it is at the lower O₂ tension buffered by leghemoglobin [7]. It would be useful to rigorously establish the relationship between the leghemoglobin binding constants for O_2 and the O_2 binding properties of the respective bacteroid oxidases, and also to compare these data with the O₂ maxima at which N₂ fixation occurs in the different symbioses. It is also pertinent to know what factors confer O2 tolerance on rhizobial species. The Azotobacter vinelandii nitrogenase is protected from O2 inactivation when complexed with an iron-sulfur protein [88], but no such mechanism has been found in Rhizobium. Brief treatment of intact soybean roots with 1 atm O₂ diminishes nitrogenase activity, but this activity is partially restored in a very short time [169], and thus an O₂ protection mechanism may also occur in legume symbioses.

IIIB. The origin of leghemoglobin heme

The prosthetic group of leghemoglobin is protoheme IX [41,64,125] and it is present in very high concentrations in legume nodules. The origin of the heme prosthetic group has been a subject of debate for some time, and three approaches have generally been taken in

addressing the question of which symbiont synthesizes heme for leghemoglobin: (i) measuring the incorporation of radioactively labelled heme precursors into heme by nodule fractions [48,70,84,85]; (ii) measuring enzyme activities of the heme biosynthesis pathway in plant and bacterial fractions [84,149,173]; and (iii) by quantitating heme in nodules formed from bacterial mutants deficient in heme biosynthesis [87,126,148,159]. δ -Aminolevulinic acid (ALA) synthase can be detected in bacteroid, but not in plant, fractions in soybean [149], serradella and lupin [84] nodules. Furthermore, the ALA synthase activity of soybean bacteroids parallels the heme concentration of developing soybean nodules [149]. ALA dehydratase activity is found in both the plant and bacteroid fractions in nodules from soybean [149], serradella and lupin [84] nodules, but the specific activity in the plant fraction decreases as the heme concentration increases in the soybean nodules [149]. These physiological data point to the bacterium as the source of leghemoglobin heme, which is easy to believe considering the large heme concentration in legume nodules and the large population of bacteria compared with the relatively few plant mitochondria in infected root cells. However, Guerinot and Chelm [87] mutated the ALA synthase structural gene (hemA) of B. japonicum, and found that soybean nodules formed from the mutant contains leghemoglobin heme. This latter finding is surprising, since most of the physiological data indicating that bacteroids synthesize leghemoglobin heme comes from work done with the soybean symbiosis [48,149,173]. It is equally interesting that a mutation in the hemeA gene of R. meliloti, from which the B. japonicum gene was derived, forms white nodules on alfalfa [126].

It is clear that neither the physiological data nor the mutant analyses of the first step in heme synthesis arrive at a unified conclusion as to which symbiont synthesizes leghemoglobin heme. More recently, a B. japonicum mutant defective in protoporphyrinogen oxidase activity, the seventh of eight enzymatic steps in heme synthesis, formed soybean nodules deficient in leghemoglobin heme [159]. Thus bacterial heme synthesis is essential for leghemoglobin expression in soybean nodules [159] even though a functional bacterial ALA synthase is not required [87]. It is important to note that nodules formed from the B. japonicum ALA synthase mutant have good nitrogenase activity as well as containing leghemoglobin heme [87]. These bacteroids must therefore contain cytochrome heme, otherwise the cells would not be able to support nitrogen fixation. It seems that the B. japonicum ALA synthase mutant can be rescued either by the plant host, or by an alternative bacterial pathway to ALA synthesis [87], whereas the protoporphyrinogen oxidase mutant cannot be rescued symbiotically [159]. It would also appear that the ALA synthase mutant of R. meliloti cannot be rescued as

alfalfa nodules formed from the mutant do not contain leghemoglobin [126].

A secondary site suppressor mutation has been described in a hemA (the structural ALA synthase gene) mutant of Azorhizobium caulinodans strain ORS571 [170] suggesting that an alternative method of ALA synthesis is possible in N_2 -fixing bacteria. The relevance of this finding, however, cannot be ascertained, since it is not yet known whether the suppressor mutant forms leghemoglobin-containing nodules on Sesbania rostrata. It is prudent not to extend these findings to other symbioses, since the hemA mutant of R. meliloti results in leghemoglobinless nodules [126], and thus an alternate pathway of ALA synthesis does not seem to function symbiotically in that bacterium.

It is possible that the B. japonicum hemA mutant was rescued symbiotically by the acquisition of ALA or by another heme precursor from the soybean host. Since the plant fraction of soybean nodules contains no measurable ALA synthase activity [149], then ALA would have to be synthesized via a different enzyme, or it would have to be made outside the nodule and then transported there. Although either of these situations could explain the apparent discrepancy in the soybean literature [87,149,159,173], neither hypothesis has been tested. It is known that plants synthesize ALA from glutamate via the five carbon pathway [21,43] rather than by ALA synthase, but the former pathway is confined to chloroplasts in eukaryotes [137]. Nevertheless, it is tempting to make an analogy between chloroplasts and infected plant roots with respect to the enormous amount of tetrapyrroles found in both structures.

Whereas arguments can be made against bacteroids as the provider of early heme precursors for leghemoglobin formation, the data thus far support a role for prokaryotic heme enzymes in catalyzing the late steps in heme biosynthesis [159,173]. Obligate aerobes use O₂ as a substrate in the two oxidative steps in heme synthesis, but the O₂ concentration in legume nodules is only about 10 nM, and thus these two enzymes, coproporphyrinogen oxidase and protoporphyrinogen oxidase, must either have a very high affinity for O_2 , or else an oxidant other than O2 is used in catalysis. Keithly and Nadler [110] found that B. japonicum bacteroids have NADP+-dependent coproporphyrinogen oxidase activity under anaerobic conditions. This activity is known to occur in anaerobes [145], but B. japonicum is the only known obligate aerobe to have 'anaerobic' coproporphyrinogen oxidase activity [110]. It is not yet known whether rhizobial species have an anaerobic protoporphyrinogen oxidase; such activity has been found in E. coli [101,102] and in Desulfovibrio gigas [112], and the D. gigas protoporphyrinogen oxidase has been purified [113]. A mutation in O₂-dependent protoporphyrinogen oxidase activity in B. japonicum results

E. coli FNR consensus	5'-A	TTGA	TATCAAT	_A-3'
B. japonicum hemA	5'-TT	TTTGAT	GGGATCAAG	- TT−3′
R. meliloti hemA	5'-TTC	CTTGACT	TCGATCGAT	GT-3'

Fig. 2. Comparison of the *E. coli* FNR consensus sequence with the *hemA* promoter of *R. meliloti* (P₂) and of *B. japonicum*. Underlined regions of the *hemA* promoters are identical to the FNR consensus sequence. See text for details.

in a heme-deficient phenotype symbiotically [159], and thus if anaerobic activity exist in nodules, it is likely to occur via the same enzyme which catalyzes the O_2 -dependent reaction.

IIIC. Regulation of heme biosynthesis

Incubation of B. japonicum cultured cells under restricted aeration results in elevated levels of ALA synthase and ALA dehydratase activities, and in the excretion of porphyrins into the growth medium [17]. These observations suggest that heme synthesis is negatively regulated by O2, which is consistent with the increased cytochrome synthesis [3,4] and possibly the leghemoglobin heme synthesis which occurs in the low O₂ environment of the nodule. In E. coli, the transcriptional activator FNR is required for the expression of numerous genes involved in anaerobic respiratory pathways, and an 'FNR consensus sequence' has been identified in the promoter regions of genes regulated by FNR [202]. The ALA synthase structural genes (hemA) of B. japonicum [139] and R. meliloti [127] have promoter sequences homologous to the FNR consensus sequence of E. coli * (Fig. 2). These findings indicate that rhizobia may contain FNR, and that it could be involved in the regulation of heme synthesis enzymes by O2. The expression of FNR in Rhizobium and Bradyrhizobium would mean that anaerobic-type regulation occurs in an obligate aerobe. It seems that rhizobia have accommodated to the very low O2 nodule environment in two ways: (i) by the expression of cytochrome oxidases which have a very high affinity for O2, thereby allowing O₂-dependent oxidative phosphorylation in nodules; and (ii) by the development of anaerobic traits such as NADP⁺-dependent coproporphyrinogen oxidase activity and possibly regulation by an FNR-type protein.

ALA synthase and ALA dehydratase activities are apparently regulated by iron as *B. japonicum* cells grown in iron-deficient media are depressed in these enzyme activities [178]. Addition of iron to extracts of iron-deficient cells does not restore ALA synthase or ALA dehydratase activities [178], and thus iron probably regulates enzyme synthesis at some level.

Heme does not seem to be a feedback inhibitor of ALA synthase in *B. japonicum*, as it is in *Rhodobacter sphaeroides* [42], since addition of heme does not repress ALA synthase activity in *B. japonicum* cells [178]. The incorporation of ALA into heme is inhibited by heme and protoporphyrin [49], but it takes $100 \, \mu \text{M}$ of porphyrin to inhibit incorporation by 50%, and thus heme may not be a physiological feedback inhibitor. Heme or protoporphyrin, however, may be a regulator of heme enzymes at the synthesis level as judged from the elevated levels of ALA synthase, ALA dehydratase and coproporphyrinogen oxidase activities in a heme synthesis mutant of *B. japonicum* deficient in protoporphyrinogen oxidase activity [159].

IIID. Heme as a regulator

As stated above, heme may regulate the synthesis of enzymes of the heme biosynthesis pathway [159], but it is not an inhibitor of these enzymes [178]. It has also been speculated that bacterial heme induces leghemoglobin apoprotein synthesis by the plant (reviewed in Refs. 7 and 35), and there is evidence that heme controls synthesis at the post-transcriptional level [236]. According to the proposed scenario, rhizobia in the developing nodule consume most of the available O₂, which in turn triggers bacterial heme synthesis. The heme then induces apoleghemoglobin formation by the plant, and thus heme and apoprotein synthesis are coordinated in the respective symbionts. However, nodules formed from a protoporphyrinogen oxidase mutant of B. japonicum contain leghemoglobin despite the heme deficiency in those nodules [159]. This indicates that the heme moiety is not an essential prerequisite for apoleghemoglobin synthesis by the plant. Also, a porphyrin-excreting mutant of R. leguminosarum forms nodules on peas which contain some apoleghemoglobin [34].

Thummler and Verma [211] found that heme binds to nodule sucrose synthase in soybean, and that heme regulates that enzyme. The sucrose synthase tetramer dissociates in the presence of heme with a concomitant inactivation of activity. It was postulated that, in the nodule, free heme accumulates during senescence when leghemoglobin breaks down, and results in the inactivation of sucrose synthase. The supply of carbon sources to bacteroids would therefore decrease during nodule senescence.

^{*} To our knowledge the similarity between the hemA promoters and the FNR consensus sequence was first brought to the attention of the scientific community in a lecture by F. de Bruijn at the 7th International Congress on Nitrogen Fixation, 13-20 March 1988, Köln, F.R.G.

IV. Electron transport

IVA. Cultured cells

Electron transport in Rhizobium has been studied most intensely in B. japonicum, and to a lesser extent in Bradyrhizobium sp. (Lupinus) and in R. leguminosarum. The complement of cytochromes expressed is significantly different in bacteroids compared with cells grown in culture. B. japonicum-cultured cells contain b- and c-type cytochromes, as well as the terminal oxidases cytochromes aa_3 and o [4]. Cytochromes o and aa_3 form complexes with carbon monoxide in a CO-O₂ atmosphere (95:5, v/v), but respiration is not completely inhibited under these conditions, and thus there is likely to be at least one other oxidase expressed in cultured cells [4]. Cytochromes c and aa_3 seem to be part of one branch of the electron-transport system, and cytochrome o is part of another branch which does not oxidize cytochrome c [4,158,160]. A mutant strain of B. japonicum which lacks cytochrome c and aa₃ has similar respiratory properties to a mutant which lacks only cytochrome aa₃ [158], suggesting that cytochromes c and aa_3 are a functional unit. Also, the isolation of B. japonicum cytochrome mutants based on a respiratory deficiency always yield lesions in the cytochromes c and aa₃ part of the electron transport system [65,158]. The electron transport system of free-living R. trifolii is similar to that of B. japonicum, with the branch point at cytochrome b, and with a- and c-type cytochromes forming a branch separate from that of cytochrome o [52]. R. trifolii [51] and R. leguminosarum [179] both express cytochrome d under restricted aeration, and although oxidase function has not been demonstrated in the cytochrome d of Rhizobium, this cytochrome is an oxidase in E. coli [75] and in other bacteria, and is expressed under O₂-limited growth in those organisms.

The expression of multiple oxidases is common in bacteria, and they allow the organism to survive in a wide range of environmental conditions. A cytochrome aa₃-deficient mutant of B. japonicum has greatly diminished NADH- and TMPD-oxidase activities, but it can oxidize succinate at wild-type levels [158]. Inhibitor data also show that NADH and TMPD are oxidized by a different oxidase than is succinate [158]. The substrate specificity may be due to the considerably higher rate of oxidation of NADH and TMPD by cell-free extracts of the parent strain compared with the oxidation rate of succinate. If this is true, it would suggest that cytochrome aa₃ has a higher turnover rate, and can support a higher electron flux than can cytochrome o. Cytochrome o, however, seems to be more suited to function under low O₂ conditions [162], and it probably has a higher affinity for O₂ than does cytochrome aa₃. Cells grown under restricted aeration have diminished, or no, cytochrome aa₃ [17,50,203], which is consistent with the idea that it is not functional under conditions of O_2 limitation.

Generally, cytochrome aa_3 and cytochrome o expression in Rhizobium is repressed in root nodules, and other oxidases unique to symbiotic growth are expressed [3,115,136,161,179]. The decrease in cytochrome aa₃ has been correlated with an increase in nitrogenase activity in B. japonicum bacteroids [45], and aa₃-type cytochrome is retained in ineffective (non-N₂-fixing) strains of Bradyrhizobium sp. (Lupinus) [136] and R. leguminosarum [116]. Despite these observations it is difficult to correlate cytochrome aa₃ and o with symbiosis, nitrogen fixation or even oxygen concentration. Cytochrome o has been observed in bacteroids from effective nodules of B. japonicum [109,161], R. leguminosarum and B. parasponia [9]. Similarly, cytochrome aa₃ is retained in effective nodule bacteroids of B. parasponia [9] and in an entire DNA homology class of B. japonicum [108], and also in cultured cells of Bradyrhizobium sp. strain 32H1 which fix nitrogen [9]. The symbiotic expression of cytochrome aa₃ in some effective B. japonicum strains [108,109] is most interesting in that these bacteria presumably function under the same leghemoglobin-buffered O₂ tension as those strains which do not express cytochrome aa₃ symbiotically, and thus oxygen cannot be the sole regulator of the expression of this oxidase. Cytochrome aa₃ seems to have a low affinity for O2 not only in Rhizobium [50,115,162], but in other bacterial strains as well [190,193,201], and so it is somewhat surprising that the oxidase is retained in bacteroids of some rhizobial strains. The presence of bacterial cytochrome aa₃ in nodules, however, does not prove that it can function at very low oxygen tensions. Oxidase activity of cytochromes aa₃ or o has not been demonstrated in nodules, and a cytochrome aa₃-deficient mutant of a B. japonicum strain which normally expresses the oxidase symbiotically still produces effective nodules despite the lesion [152].

IVB. Bacteroids

Bradyrhizobium japonicum bacteroids express cytochromes c-552, c-554, P-420, and P-450, all of which react with carbon monoxide and are thus putative oxidases [3,220]. Bradyrhizobium sp. (Lupinus) bacteroids also express cytochrome P-450 [136] as do, to a lesser extent, R. leguminosarum strain PRE and cultured cells of Bradyrhizobium sp. strain 32H1 induced for nitrogenase activity [9]. B. japonicum bacteroids also contain a non-heme component which has high cytochrome c oxidase activity; it is CO-insensitive and may be a flavin-containing metalloprotein [6,161].

Appleby [3] was unable to demonstrate that the CO-reactive proteins found in B. japonicum bacteroids

were cytochrome oxidases by photochemical action spectra; it was later discovered [29] that bacteroid respiration is insensitive to CO at the O₂ concentration (70 μ M) at which the action spectra were performed. Carbon monoxide inhibits respiration at leghemoglobin-buffered O_7 concentrations (less than 0.1 μ M; see Ref. 29), but it is not possible to maintain such a low O₂ concentration experimentally for performing action spectra without introducing an O₂ carrier protein such as leghemoglobin or myoglobin, which have their own CO-binding properties. Thus, elucidation of bacteroid electron transport has proceeded by other means, and proof of oxidase function of the numerous CO-reactive proteins has not yet occurred. Bergersen and Turner [30] identified four O_2 affinity states of B. japonicum respiration, which presumably correspond to four oxidases or to four O2 binding sites with different affinities for O_2 . The highest O_2 affinity state showed an apparent K_s value of 6 nM [30], which is close to the estimated O_2 concentration of 10 nM within a soybean nodule [5,213]. Also, hydrogen oxidation by B. japonicum bacteroid membranes involved three cyanide-binding sites [161], supporting the CO-binding data and the respiration kinetics experiments for multiple oxidases. The multiple oxidase system of B. japonicum bacteroids apparently allows efficient and inefficient electron transport within the nodule with respect to ATP synthesis. The addition of leghemoglobin to bacteroid suspensions resulted in a large increase in nitrogenase activity with only a modest increase in O₂ consumption [31,232], indicating that the ATP/O ratio is greater at the low O₂ concentration that leghemoglobin maintains. It was directly shown that ATP synthesis is highest in B. japonicum bacteroids at leghemoglobin-buffered O2 concentrations of 0.02-0.1 μ M [29], showing that efficient respiration occurs at a low O₂ tension. This efficient phase of respiration is inhibited by carbon monoxide and N-phenylimidazole, whereas the less efficient respiration that occurs above 1 μ M O₂ is insensitive to these inhibitors [13,29,30]. Nphenylimidazole has been shown to be an inhibitor of cytochrome P-450 [13], and thus this cytochrome, although perhaps not an oxidase, is essential for the efficient phase of respiration which supports nitrogen fixation.

At high O_2 concentrations (more than 1 μ M) B. japonicum bacteroid respiration is uncoupled from ATP synthesis [29] and does not involve cytochrome P-450 as judged from inhibitor data [13,29,30]. It has been proposed that the uncoupled electron transport acts as 'protective respiration' to maintain a low intracellular O_2 concentration at the expense of high-energy electrons to prevent inactivation of the O_2 -labile nitrogenase enzyme [29]. A protective respiration mechanism has been proposed for Azotobacter (see Ref. 174) which is an obligate aerobe with a very high respiratory rate and with no known hemoglobin-type protein to buffer the

O₂ concentration. In these systems, electron flow through the cytochrome $b \rightarrow d$ branch is very rapid, but energy is not conserved by that pathway [57]. Appleby [6] discovered a very vigorous cytochrome c oxidase activity in B. japonicum which is sensitive to cyanide and other metal-chelating agents but not to carbon monoxide. The high turnover rate and the CO-insensitivity of this activity led to the conclusion that the oxidase terminated the inefficient branch of bacteroid electron transport [6]. Bacteroid cytochrome c oxidase activity is inhibited by Atebrin [6,161] which can be partially reversed by FMN or FAD [6], suggesting that the oxidase may be a flavoprotein. Also, H2-reduced bacteroid membranes show a flavin absorption spectrum [161] that disappears in the presence of cyanide; the membrane-bound hydrogenase of B. japonicum is not a flavin-containing dehydrogenase, and thus the flavin must be due to another component. The insensitivity to CO suggested that the cytochrome c oxidase may be a non-heme metalloprotein [6]; this speculation was strengthened by the recent observation that a cytochrome-deficient mutant of B. japonicum retained almost wild-type levels of bacteroid cytochrome c oxidase activity [158]. The mutant analyses also show that bacteroid cytochrome c oxidase can be expressed independently of cytochrome c [158], even though cytochrome c is likely to be the physiological substrate of the oxidase [6].

A cytochrome aa₃-deficient mutant yielded bacteroids with elevated cytochrome c oxidase activity [158] which suggests that, even though the two oxidases are not expressed simultaneously, their regulation is somewhat related. Alternatively, protective respiration, and hence bacteroid cytochrome c oxidase, may only be necessary in young nodules where the O_2 concentration would be relatively high [7], and the cytochrome aa₃ mutation may result in less developed nodules with higher bacteroid oxidase activity. It should be noted, however, that those nodules formed from the cytochrome aa₃ mutant appear to be normal in several respects. The wild-type gene which was mutated to yield the cytochrome aa₃-deficient phenotype has been cloned [164]; it codes for a diffusible product that complements the cytochrome aa₃ mutant. It is not yet known if this gene product represses bacteroid cytochrome c oxidase in the mutant to wild-type levels.

V. ATP synthesis and ion transport

The most accepted notion of how ATP is synthesized in bacteria, as well as in mitochondria and chloroplasts, is by the coupling of H^+ diffusion in the direction of an electrochemical gradient with the phosphorylation of ADP. The electrochemical gradient, or protonmotive force, is the sum of the electrical potential across the bacterial membrane ($\Delta\Psi$) and the proton chemical

gradient between the internal and external pH (Δ pH). $\Delta\Psi$ is most commonly measured indirectly by quantitating the distribution of lipophilic ions (or ions made membrane-permeable by the addition of an ionophore) on both sides of the membrane, and calculating the membrane potential using the Nernst equation with the assumption that the ions are in equilibrium (see Ref. 104). Another model for ATP synthesis abandons the need for a potential across the membrane in favor of a localized proton current where H⁺ diffusion is controlled at the level of the ATPase rather than at the membrane [228]. The latter model does not mandate a fixed correlation between H⁺ flux and ATP synthesis, and thus the relationship between ATP/O ratio and H⁺/O measurements cannot be assumed. Also, the interpretations of permeant ion distribution measurements inside and outside a cell as an indicator of its energy status can only be valid in terms of a protonmotive force as defined above for chemiosmosis. Although we do not intend to refute the chemiosmotic hypothesis here, it is our opinion that the alternate model described above [228], and also the criticisms raised against indirect measurements of $\Delta\Psi$ [209,228] cannot be ignored out-of-hand.

VA. ATP synthesis in N2-fixing cells

ATP synthesis in B. japonicum bacteroids is most efficient at a low, leghemoglobin-buffered O₂ concentration, and the ATP/ADP ratio decreases as the O₂ tension is raised [29]. The control of ATP synthesis by O_2 in N_2 -fixing organisms has been explained in terms of a branched electron-transport system with the ATPefficient branch terminated with a high O₂-affinity oxidase, and with the inefficient branch terminated by an oxidase with a lesser affinity for O₂ [29,30]. However, high ATP/ADP ratios occur in R. leguminosarum bacteroids [120] at an O_2 concentration (approx. 1 μ M) where respiration is uncoupled in B. japonicum [29]. Similarly, the ATP/ADP ratio increases markedly in R. phaseoli bacteroids when the O₂ concentration is raised from 3 to 7 μ M [215]. Experiments with R. leguminosarum [120,121] and B. japonicum [29] show that ATP synthesis increases as a function of the O₂ concentration at very low O₂ tensions, but the optimal O₂ concentration for ATP synthesis and nitrogen fixation are quite different in the two systems. The different O2 optima for ATP synthesis could be due to oxidases with different affinities for O_2 , but the cytochrome profile of lupin bacteroids [136] looks very similar to that of soybean bacteroids [3], yet ATP synthesis is optimal at a higher O₂ concentration in lupin bacteroids compared with that of soybean bacteroids [7].

Appleby et al. [13] found a linear relationship between ATP/ADP ratio and nitrogenase activity in B. japonicum bacteroids and concluded that the ATP/ADP

ratio controls that activity. Having established that the ATP/ADP ratio is dependent on the O_2 concentration [29,30], then nitrogenase activity is controlled by O_2 via ATP synthesis. However, Laane et al. reported that nitrogenase activity decreased with increasing O₂ without a concomitant decrease in the ATP/ADP ratio in R. leguminosarum [120]. They correlated a decrease in nitrogenase activity with a valinomycin- or nigericin-dependent decrease in the membrane potential, as measured by the distribution of tetraphenylphosphonium ion (TPP+), and concluded that $\Delta\Psi$, independently of the protonmotive force, regulates nitrogenase activity [122]. Since the ATP/ADP ratio remained constant in these experiments, they assumed that $\Delta\Psi$ regulates nitrogenases by controlling the supply of electrons to the enzyme complex.

There are several other examples where the relationship between ATP synthesis, nitrogen fixation activity and protonmotive force measurements is not obvious. Trinchant et al. [215] determined that the ATP/ADP ratio increases under conditions where nitrogenase is suppressed in R. phaseoli bacteroids. The protonmotive force, as inferred by the distribution of benzoic acid and TPP+ to deduce ΔpH and $\Delta \Psi$, respectively, of B. japonicum was somewhat less in N2-fixing bacteroids than in cultured cells [32]. Although this observation can be accounted for by the assumption that the assay was performed at high O2 tensions where N2 fixation is not likely to occur in either cell type, it should be noted that the electrochemical potential only decreased from -173 mV to -136 mV when 2.5 mM azide was added [32], and thus the protonmotive force is not severely affected by cell metabolism. Gober and Kashket [81] compared Bradyrhizobium sp. strain 32H1 grown in air and under 0.2% O_2 to derepress nitrogenase activity. The protonmotive force of the air-grown cells was -237mV at pH 6.1, but it was only -115 mV in the N_2 -fixing cells when assayed at 21% or 0.2% O_2 . Despite the higher protonmotive force in the air-grown cells, the intracellular ATP concentration was higher in the N₂fixing cells at pH 6.1, and the ATP/ADP ratio was about the same under both growth conditions. These findings are different from those of Laane et al. [120-122] in that the decrease in the membrane potential does not seem to shut off nitrogenase activity in Bradyrhizobium sp. strain 32H1 [81] as it does in R. leguminosarum [120,122]. Gober and Kashket [81] offered several hypotheses for the low membrane potential and protonmotive force in the N₂-fixing cells. One idea is that the cells grown under 0.2% O₂ (N₂-fixing) have a 'leakier' membrane with respect to protons than do the air-grown cells resulting in a collapse of the membrane potential. Subsequently, they found that a K⁺/H⁺ antiporter is induced at 0.2% O₂, which was credited with collapsing the membrane potential [82]. They also proposed that ATP synthesis may be driven

by localized proton currents as proposed by Williams [227,228], and thus the protonmotive force "does not reflect the true driving force for ATP synthesis" [81]. Indeed, the measured protonmotive force does not seem to correlate with the ATP concentration or the ATP/ADP ratio inside the cell. In those experiments, growth under 0.2% O_2 would result in an alteration in the proton circuit which drives ATP synthesis [81]. The induction of the K^+/H^+ antiporter would certainly alter the proton current, but under a chemiosmotic scheme the antiporter should have an uncoupling effect.

VB. Ion transport

VB-1. Potassium

A decrease in the membrane potential of Bradyrhizobium sp. strain 32H1 grown under 0.2% O2 compared with air-grown cells led Gober and Kashket [81] to speculate that a H⁺-linked flux of another ion occurs in the cells grown under low $[O_2]$. They found a K^+/H^+ antiporter that required high [K+] and low [O2] during growth for induction [82]. The antiporter was detected from the collapse of $\Delta\Psi$ when induced cells were-loaded with K⁺ and put in a low [K⁺] medium; this treatment had no effect on the $\Delta\Psi$ in air-grown cells or in cells grown in low [K+]. The K+/H+ antiporter was also discerned by a K⁺-dependent increase in the proton conductance of the membrane of cells grown under low [O₂], high [K⁺]. K⁺ did not alter H⁺ conductance in cells grown under high [O₂] and/or low K⁺, and Na⁺ did not increase H⁺ conductance in cells where the antiporter is supposed to be expressed. This antiporter system is different from the K+-ATPase of Rhizobium sp. strain UMKL20 [129] in that the latter is induced by 50 μM K⁺ in air. The effects of K⁺ in Bradyrhizobium also differ from other bacteria in that it alters the protonmotive force as well as the membrane potential in Bradyrhizobium sp. strain 32H1 [82]. In several other bacteria [19,91,107,116,117,150,200,214] a K+-induced decrease in $\Delta\Psi$ is accompanied by an increase in the Δ pH, and thus the magnitude of the protonmotive force is not significantly changed. The K⁺/H⁺ antiporter-dependent decrease in the protonmotive force in Bradyrhizobium sp. strain 32H1 is not accompanied by a decrease in ATP synthesis. Cells grown under conditions that induce the K⁺/H⁺ antiporter have nitrogenase activity, and have a higher or comparable intracellular ATP concentration and ATP/ADP ratio compared with cells repressed for the antiporter [81].

The effect of K^+ on cultured cells of *Bradyrhizobium* sp. strain 32H1, and also of *B. japonicum*, apparently have significance in the symbiosis as growth in K^+ and 0.2% O_2 resulted in traits normally associated with bacteroids. Cells grown in 8 mM K^+ and 0.2% O_2 expressed nitrogenase activity and an ammonium transport carrier, increased expression of δ -aminolevulinic

acid (ALA) synthase and ALA dehydratase activities, and repression of glutamine synthetase activity and of capsular polysaccharide synthesis [83].

VB-2. Molybdenum

Molybdenum is a component of the nitrogenase enzyme and, since it is a trace metal, it is reasonable to assume that N₂-fixing organisms have a mechanism for acquiring molybdenum from the environment. Klebsiella pneumoniae has a high affinity molybdenum uptake system that is constitutively expressed and is not energy-dependent [100,197], and Clostridium pasteurianum and Azotobacter vinelandii can store molybdenum in a protein-associated form [146,172]. Molybdenum uptake in B. japonicum cells induced for nitrogenase activity in culture showed biphasic kinetics indicative of a high- and a low-affinity system for molybdenum influx [133]. Mutants in molybdenum metabolism were isolated based on their ability to grow on nitrate only in the presence of added molybdenum. Two mutants were isolated which lack the high-affinity uptake system. Molybdenum uptake and nitrogenase activity were restored in these mutants by addition of spent media in which wild-type cells had grown. These data indicate that B. japonicum excretes an extracellular molybdenum-binding factor involved in the high-affinity uptake system. Nitrogenase activity was also restored in the molybdenum uptake mutants by the addition of 10-100 mM molybdenum to the depression medium [133]. The ability to accumulate molybdenum varies among naturally occurring strains of B. japonicum [86]; the variability can be accounted for by the differing affinities of the molybdenum uptake systems for molybdenum between the strains examined [86].

VB-3. Nickel

Hydrogenase is a nickel enzyme in B. japonicum (subsection VIB), and it must acquire this metal from the environment. Mutant strains of B. japonicum which express hydrogenase constitutively accumulate Ni2+ at a much higher rate than does the parent strain [205]. The K_m of Ni^{2+} uptake for Ni^{2+} is similar in both cell types, but the V_{max} is much higher in the hydrogenase constitutive strains. Ni²⁺ uptake is inhibited significantly by Zn²⁺ and by Cu²⁺, and to a lesser extent by Co²⁺, Mg²⁺ and Mn²⁺. The nickel uptake system in B. japonicum, therefore, does not seem to occur via a Mg²⁺ uptake system as occurs in some other bacteria [106]. Nickel uptake is believed to be energy-dependent in Methanobacterium bryantii [103], but not in Azotobacter chroococcum [168]. The addition of respiratory inhibitors or uncouplers of ATP synthesis has little effect on Ni²⁺ uptake in cultured cells of B. japonicum [205], and thus influx of this metal is not energy-dependent.

VB-4. Ammonium

Ammonium-ion transport is obviously important in symbiotic nitrogen-fixers, since fixed nitrogen is exported to the host plant. Methylammonium is often used as an ammonium analog, and in cases where NH₄⁺ has been shown to compete with methylammonium, the use of the analog is valid [80,123]. There is also a methylammonia permease that does not transport NH₄⁺ [77]. The ammonium uptake system is induced under N-limited conditions in R. leguminosarum [77] and in Bradyrhizobium sp. strain 32H1 under conditions where nitrogenase activity is induced [80,83]. Dilworth and Glenn [55] could not find an ammonium uptake system in R. leguminosarum bacteroids, and thus argued that this system is probably important for the accumulation of NH₄ in the soil, but not in the nodule. The induction of the methylammonium transport system in Bradyrhizobium sp. strain 32H1, however, coincides with the expression of numerous traits associated with bacteroids [83], which hints at the idea that this transport system has significance symbiotically. The methylammonium uptake system is energy-dependent as seen by the inhibition of influx by metabolic inhibitors and uncouplers [80]. R. leguminosarum bacteroids loaded with methylammonium effluxed the NH₄⁺ analog in response to O₂ pulses [123], indicating that NH₄⁺ export is also energy-dependent in this bacterium.

VI. Hydrogen oxidation

A hydrogen uptake hydrogenase has been identified in some strains of B. japonicum, Bradyrhizobium sp. (mungbean and cowpea), R. meliloti (very poor activity), R. leguminosarum [38], and in the stem-nodulating bacterium Azorhizobium caulinodans strain ORS571 [203]. The source of H₂ for hydrogenase is from the nitrogenase enzyme; H⁺ is reduced to H₂ as N₂ is fixed to NH₃. H₂ evolution by nitrogenase is obligatory, and it cannot be eliminated even under experimental circumstances which heavily favor N₂ reduction over H₂ production [177,199]. Under these conditions approx. 25% of the energy required for nitrogen fixation is committed to H₂ production [199], and under physiological conditions in root nodules 40-60% of the energy input to nitrogen fixation is 'lost' to H₂ evolution [195]. Thermodynamic considerations mandate that the energy obtained from H₂ oxidation cannot completely compensate for the cost of H₂ production, but it is plausible that H₂ oxidation via hydrogenase yields energy that would otherwise not be available to the cell [56]. The hydrogen uptake (Hup) phenotype of B. japonicum has been credited with conferring superior traits on soybean [2,68], and with maintaining a higher steady-state intracellular ATP level in isolated bacteroids [66]. Soybeans grown in an aerated liquid medium, however, showed no differences when inoculated with a

Hup mutant of B. japonicum compared with those inoculated with a Hup+ revertant strain [59]. Those authors argued that O₂ is limiting and thus the alleged lower ATP/O ratio with hydrogen as the substrate compared with carbon sources has a deleterious effect on the energy status of the cell. Other studies have shown that varying the amount of O₂ available to soybean roots does not have a long-term effect on nitrogen fixation [46,144], indicating that O₂ is not limiting in nodules. For the most part, the literature addressing the R. leguminosarum hydrogenase does not make a convincing case for H2 oxidation being a beneficial trait in this organism. Only 5 of 14 Hup+ strains of R. leguminosarum show 'H2-dependent ATP synthesis [153] in isolated bacteroids, and hydrogenase activity does not increase symbiotic nitrogen fixation [47,218].

VIA. Regulation

Hydrogenase expression in Rhizobium is regulated by factors associated with its symbiosis with the plant, and cultured cells do not normally synthesize the enzyme. These factors not only include plant compounds, but also the nodule environment itself, since hydrogenase activity can be induced in culture under certain conditions [132,134,141,143]. To a certain extent hydrogenase derepression conditions mimic the nodule environment, and regulation at this level is much better understood than regulation by specific plant products. It has long been thought that organic carbon represses ex planta hydrogenase expression in B. japonicum [134], but recent experiments show that hydrogenase can be expressed when cells are grown heterotrophically in the presence of H₂ and under low O₂ concentrations [222]. Data indicate that carbon repression of hydrogenase in culture is probably caused by acidification and O₂ depletion of the growth medium, and precautions taken to buffer the cultures relieves repression by organic carbon substrates [224]. However, spontaneous mutants of B. japonicum which express hydrogenase constitutively also show constitutive expression of ribulose-1,5-bisphosphate carboxylase activity [142], and thus the regulation of hydrogenase and carbon metabolism are lin-

Hydrogenase expression is regulated by O_2 , and a low O_2 tension is required for activity in cultured cells [132,134]. *B. japonicum* mutants have been obtained which are either hypersensitive [135] or insensitive [141,143] to O_2 . The nature of the genetic lesions in the O_2 regulatory mutants are not yet known, nor is it known how O_2 regulates hydrogenase expression. However, Novak and Maier [157] found that *B. japonicum* hydrogenase activity is repressed in cultured cells when the DNA gyrase inhibitors coumermycin, novobiocin or nalidixic acid are added to the derepression medium. These repressed cells do not synthesize the hydrogenase

subunits, but protein synthesis in general is not affected [157]. Hydrogenase is therefore regulated by DNA topology; DNA gyrase activity has been shown to affect bacterial genes that are regulated by O₂ [18,114,233], including nitrogen fixation genes in *Rhodobacter capsulatus* and *Klebsiella pneumoniae* [114]. Hydrogenase-constitutive mutants of *B. japonicum* do not appear to have greater gyrase activity than does the wild type, but hydrogenase synthesis is not repressed by gyrase inhibitors in the constitutive strains [156].

The expression of hydrogenase is also regulated by the plant host in a manner which is poorly understood. Several B. japonicum strains have hydrogenase activity in cowpea (Vigna unguiculata) nodules, but have no activity in symbiosis with soybean [111]. Bacterial protein (B. japonicum strain USDA 61) from those Hup⁺ cowpea nodules do not cross-react with antihydrogenase antibody from B. japonicum strain SR [223] which suggests either that the enzyme from the 'host control' strain is antigenically different from the conventional enzyme, or that the hydrogenase protein concentration was too low to be detected antigenically. Host control of hydrogenase has also been observed between different cultivars of the same pea (Pisum sativum) strain [22-24]. Grafting experiments showed that a transmissible shoot factor controls the expression of hydrogenase activity in the different pea cultivars [23]. The grafting data also showed that the plant host affected the H_2/N_2 ratio in nodules formed from a Hup-strain, and thus the host can influence the efficiency of nitrogen fixation [23]. The fact that high electron flux through purified nitrogenase favors N₂ reduction over H₂ production [89] leaves open the possibility that host control of H₂ metabolism is at the level of photosynthate supply to the nodule bacteroids.

Nickel is part of the hydrogenase enzyme in B. japonicum (subsection VIB), and it has been implicated as a regulator of hydrogenase expression [208]. Hydrogenase protein and activity both increase as a function of nickel concentration; this enhancement can be eliminated by the addition of rifampicin, which suggests that nickel regulates hydrogenase at the transcriptional level [208]. A similar phenomenon also occurs in Alcaligenes latus [58], but nickel is not required for hydrogenase apoprotein synthesis in Methylosinus trichosporium [44] or for carbon monoxide dehydrogenase apoprotein synthesis in Rhodospirillum rubrum [37]. In those nickel enzymes not regulated by the metal, enzyme activity can be demonstrated by the addition of nickel to the pre-formed protein.

VIB. Components of H2 oxidation

The *B. japonicum* hydrogenase enzyme has been purified from bacteroids [14], heterotrophically grown cells [206,207], and chemolithotrophically grown cells

[90]. The enzyme is a dimer with subunits of approx. 65 kDa and 33-35 kDa in size, and it contains nickel. Nickel has been found in hydrogenases from many organisms, but the actual role of the metal is not known for certain. H₂ can react with nickel to form nickel hydrides [210], and H₂ apparently binds to nickel in the hydrogenase of Desulfovibrio gigas [124]. However, the hydrogenase of Clostridium pasteurianum [1] and D. vulgaris [97] are very active, yet they do not contain nickel, and thus it is premature to make generalizations as to the specific role of nickel in the B. japonicum hydrogenase.

Although hydrogenases are treated as O2-labile enzymes, the B. japonicum enzyme has been successfully purified aerobically by affinity chromatography, and these preparations have superior methylene blue-dependent H₂ uptake activity [206]. B. japonicum hydrogenase purified anaerobically can be inactivated by O₂ in both a reversible and irreversible manner [196]. The reversible inactivation requires the presence of an electron acceptor from hydrogenase and high pH [196]. Moshiri and Maier [147] found that B. japonicum hydrogenase in membrane vesicles is susceptible to proteases and membrane-impermeant protein modification reagents only when the enzyme is reduced, and not in the oxidized form. The inactivation of activity by these compounds is attributed specifically to the degradation or modification of the approx. 33 kDa subunit [147]. These data indicate that hydrogenase undergoes a conformational change in the membrane which is redox-dependent.

Hydrogen oxidation yields energy via electron transport, and cytochromes and ubiquinone has been directly [160-163] and indirectly [61-63] identified which are reducible by H₂ in B. japonicum cells under various growth conditions. In addition, the Evans laboratory has claimed that there is a low-potential b-type cytochrome, called component 559-H2, in bacteroids and in chemolithotrophically grown cells that is specifically involved in H₂ oxidation, but not in the oxidation of other substrates [62,63]. O'Brian and Maier have provided evidence that this component does not exist in bacteroids [165] or in hydrogenase-constitutive cultured cells possessing vigorous hydrogen uptake activity [162,163]. Furthermore, it is the reviewers' opinion that the experimental evidence provided by the Evans group [62,63] in support of component 559-H₂ cannot withstand scrutiny, and that their conclusions are actually contradictory to their own data. Firstly, the b-type cytochrome assigned 'component 559-H₂' [62] is not solely reducible by H2, but also by endogenous substrate [62,63], and thus it is not unique to H₂ oxidation. The fact that some cytochrome b is reduced faster by H₂ plus endogenous substrate than by endogenous substrate alone is not relevant, since this was found to occur with the c-type cytochrome as well [63]. Secondly,

the claim that component 559-H₂ has a low redox potential and accepts electrons from hydrogenase [62] is contrary to the observation that the component remains oxidized when other cytochromes alleged to be part of the same branch of the electron-transport system are reduced [62,63]. Thirdly, the assertion that component 559-H₂ is present in bacteroids [62] is inconsistent with the data of Eisbrenner and Evans [61] which show this component to be absent in those cells. Finally, the method of quantifying component 559-H₂ must be incorrect, since Eisbrenner and Evans claim to measure H₂-dependent component 559-H₂ reduction spectrophotometrically in B. japonicum strains which cannot use H₂ as a reductant [62]. In a recent review, Evans et al. [69] recalculated the correlation between hydrogenase activity and component 559-H₂ expression (from Ref. 62) by omitting the data from Hup strains, and they still found a high correlation between the two parameters. This correction, however, does not rectify the error in the method of quantifying component 559-H₂, and thus the new correlation data are still meaning-

VIC. Genetics

Genes involved in H₂ oxidation are plasmid-borne in R. leguminosarum, and are co-transferred with nodulation genes [39]. B. japonicum hup genes are presumably on the chromosome as no plasmids have been identified in Hup⁺ strains of this organism. The structural genes of the B. japonicum hydrogenase have been identified and subcloned from a 31 kb DNA fragment which complements Hup⁻ mutants [234]. DNA fragments of B. japonicum containing the hydrogenase structural genes have been used to complement hup⁻ mutants of R. leguminosarum [105], and have also been used as a probe to isolate R. leguminosarum hup genes [128]. The hydrogenases of B. japonicum and R. leguminosarum are therefore genetically similar, and may have been derived from a common ancestor.

In addition to the structural genes, mutant analyses indicate that there are other genes essential for H₂ oxidation in B. japonicum. Hom et al. [93] complemented a B. japonicum point mutant deficient in both hydrogenase and nitrogenase activities with an approx. 23 kb DNA fragment cloned into a broad host range vector. This recombinant plasmid, pSH22, is very likely homologous to pHU1, which contains the hydrogenase structural genes [234], as judged from the similar restriction maps [92,94] and from the fact that the hydrogenase structural genes obtained from a gene expression library hybridize to pSH22 (Moshiri and Maier, unpublished results). Hup mutants not complemented by pSH22 have been isolated [93,94], and thus there are hup-related genes outside the DNA region contained on pSH22. The regulation of hup genes at the molecular

level is an important topic of which almost nothing is known. Mutants have been isolated which affect the regulation of Hup expression [131,135,141-143] whose defective gene has not been located or characterized.

Note added in proof (received 28 March 1989)

Since the acceptance of this review, numerous important findings have been published which pertain to *rhizobial* heme and leghemoglobin [237–240], hydrogen oxidation [241], ion transport [242] and energetics [243]. In addition, the proceedings of the Seventh International Congress on Nitrogen Fixation have been published in book form [244].

Acknowledgements

This manuscript is dedicated to the memory of Barry Chelm. The authors thank Drs. Clive Ronson and Turlough Finan for providing preprints of their manuscripts. Work from the laboratory of Robert Maier was supported by the United States Department of Agriculture, Allied Corporation and the National Institutes of Health.

References

- 1 Adams, M.W.W. and Mortenson, L.E. (1984) J. Biol. Chem. 259, 7045-7055.
- 2 Albrecht, S.L., Maier, R.J., Hanus, F.J., Russell, S.A. and Evans, H.J. (1979) Science 203, 1255-1257.
- 3 Appleby, C.A. (1969) Biochim. Biophys. Acta 172, 71-87.
- 4 Appleby, C.A. (1969) Biochim. Biophys. Acta 172, 88-105.
- 5 Appleby, C.A. (1969) Biochim. Biophys. Acta 188, 222-229.
- 6 Appleby, C.A. (1978) in Functions of Alternative Oxidases (Degn. H., Lloyd, D. and Hill, G., eds.), pp. 11-20.
- 7 Appleby, C.A. (1984) Annu. Rev. Plant Physiol. 35, 443-478.
- 8 Appleby, C.A. (1984) in Nitrogen Fixation and CO₂ Metabolism (Ludden, P.W. and Burris, J.E., eds.), pp. 41-51, Elsevier Science Publishing Co., Inc., New York.
- 9 Appleby, C.A., Bergersen, F.J., Ching, T.M., Gibson, A.H., Gresshoff, P.M. and Trinick, M.J. (1981) in Current Perspectives in Nitrogen Fixation (Gibson, A.H. and Newton, W.E., eds.), p. 369, Australian Academy of Sciences, Canberra.
- 10 Appleby, C.A., Bogusz, D., Dennis, E.S., Fleming, A.I., Landsman, J. and Peacock, W.J. (1988) 7th International Congress on Nitrogen Fixation, 13-20 March, Cologne, Abstr. L-11.5.
- 11 Appleby, C.A., Bradbury, J.H., Morris, R.J., Wittenberg, B.A., Wittenberg, J.B. and Wright, P.E. (1983) J. Biol. Chem. 258, 2254-2259.
- 12 Appleby, C.A., Tjepkema, J.D. and Trinick, M.J. (1983) Science 220, 951-953.
- 13 Appleby, C.A., Turner, G.L. and Macnicol, P.K. (1975) Biochim. Biophys. Acta 387, 461-474.
- 14 Arp, D.J. (1985) Arch. Biochem. Biophys. 237, 504-512.
- Arwas, R., Glenn, A.R., McKay, I.A. and Dilworth, M.J. (1986)
 J. Gen. Microbiol. 132, 2743-2747.
- 16 Arwas, R., McKay, I.A., Rowney, F.R.P., Dilworth, M.J. and Glenn, A.R. (1985) J. Gen. Microbiol. 131, 2059-2066.
- 17 Avissar, Y.J. and Nadler, K.D. (1978) J. Bacteriol. 135, 782-789.
- 18 Axley, M.J. and Stadtman, T.C. (1988) Proc. Natl. Acad. Sci. USA 85, 1023-1027.

- 19 Bakker, E.P. and Mangerich, W.E. (1981) J. Bacteriol. 147,
- 20 Baulcombe, D. and Verma, D.P.S. (1978) Nucl. Acids Res. 5, 4141-4153.
- 21 Beale, S.I., Gough, S.P. and Granick, S. (1975) Proc. Natl. Acad. Sci. USA 72, 2719-2723.
- 22 Bedmar, E.J., Edie, S.A. and Phillips, D.A. (1983) Plant Physiol. 72, 1011-1015.
- 23 Bedmar, E.J. and Phillips, D.A. (1984) Plant Physiol. 75, 629-633.
- 24 Bedmar, E.J. and Phillips, D.A. (1984) Can. J. Bot. 62, 1682-1686.
- 25 Bergersen, F.J. (1965) Aust. J. Biol. Sci. 18, 1-9.
- 26 Bergersen, F.J. and Appleby, C.A. (1981) Planta 152, 534-543.
- 27 Bergersen, F.J. and Turner, G.L. (1967) Biochim. Biophys. Acta 141, 507-515.
- 28 Bergersen, F.J. and Turner, G.L. (1975) J. Gen. Microbiol. 89, 31-47
- 29 Bergersen, F.J. and Turner, G.L. (1975) J. Gen. Microbiol. 91, 345-354.
- 30 Bergersen, F.J. and Turner, G.L. (1980) J. Gen. Microbiol. 118, 235-252.
- 31 Bergersen, F.J., Turner, G.L. and Appleby, C.A. (1973) Biochim. Biophys. Acta 292, 271–282.
- 32 Bhandari, B. and Nicholas, D.J.D. (1985) J. Bacteriol. 164, 1383-1385.
- 33 Birkenhead, K., Manian, S.S. and O'Gara, F. (1988) J. Bacteriol. 170, 184-189.
- 34 Bisseling, T., Been, C., Klugkist, J., Van Kammen, A. and Nadler, K. (1983) EMBO J. 2, 961-966.
- 35 Bisseling, T., Van den Bos, R.C. and Van Kammen, A. (1986) in Nitrogen Fixation, Vol. 4: Molecular Biology (Broughton, W.J. and Puhler, S., eds.), pp. 280-312, Oxford University Press, New York.
- 36 Bolton, E., Higgisson, B., Harrington, A. and O'Gara, F. (1986) Arch. Microbiol. 144, 142-146.
- 37 Bonam, D. and Ludden, P.W. (1987) J. Biol. Chem. 262, 2980–2987.
- 38 Brewin, N.J. (1984) in Genes Involved in Plant-Microbe Interactions (Verma, D.P.S. and Hohn, T., eds.), pp. 179-203, Springer-Verlag, New York.
- 39 Brewin, N.J., DeJong, T.M., Phillips, D.A. and Johnston, A.W.B. (1980) Nature (London) 288, 77-79.
- 40 Brill, W.J. (1980) Microbiol. Rev. 44, 449-467.
- 41 Broughton, W.J., Dilworth, M.J. and Godfrey, C.A. (1972) Biochem. J. 127, 309-314.
- 42 Burnham, B.F. and Lascelles, J. (1963) Biochem. J. 87, 462-472.
- 43 Castelfranco, P.A. and Beale, S.I. (1983) Annu. Rev. Plant Physiol. 34, 241-278.
- 44 Chen, Y.-P. and Yoch, D.C. (1987) J. Bacteriol. 169, 4778-4783.
- 45 Ching, T.M., Hedtke, S. and Newcomb, W. (1977) Plant Physiol. 60, 771-774.
- 46 Criswell, J.G., Havelka, U.D., Quebedeux, B. and Hardy, R.W.F. (1976) Plant Physiol. 58, 622-625.
- 47 Cunningham, S.D., Kapulnik, Y., Brewin, N.J. and Phillips, D.A. (1985) Appl. Environ. Microbiol. 50, 791-794.
- 48 Cutting, J.A. and Schulman, H.M. (1969) Biochim. Biophys. Acta 192, 486–493.
- 49 Cutting, J.A. and Schulman, H.M. (1972) Biochim. Biophys. Acta 229, 58-62.
- 50 Daniel, R.M. and Appleby, C.A. (1972) Biochim. Biophys. Acta 275, 347-354.
- 51 De Hollander, J.A. (1981) Ph.D. Thesis, Vrije Universiteit, Amsterdam.
- 52 De Hollander, J.A. and Stouthamer, A.H. (1980) Eur. J. Biochem. 111, 473-478.
- 53 Dilworth, M.J. and Appleby, C.A. (1979) in A Treatise on Dinitrogen Fixation, Vol. 2 (Hardy, R.W.F., Bottomley, F. and Burns, R.C., eds.), pp. 691-764, Wiley & Sons, New York.

- 54 Dilworth, M.J., Arwas, R., McKay, I.A., Saroso, S. and Glenn, A.R. (1986) J. Gen. Microbiol. 132, 2733-2742.
- 55 Dilworth, M.J. and Glenn, A.R. (1985) in Nitrogen Fixation and CO₂ Metabolism, (Ludden, P.W. and Burris, J.E., eds.), pp. 53-61, Elsevier Science Publishing Co., Inc., New York.
- 56 Dixon, R.O.D. (1972) Arch. Microbiol. 85, 193-201.
- 57 Downs, A.J. and Jones, C.W. (1975) FEBS Lett. 60, 42-46.
- 58 Doyle, C.M. and Arp. D.J. (1988) J. Bacteriol. 170, 3891-3896.
- 59 Drevon, J.J., Kalia, V.C., Heckmann, M.O. and Salsac, L. (1987) Appl. Environ. Microbiol. 53, 610-612.
- 60 Dreyfus, B.L., Elmerich, C. and Dommergues, Y.R. (1983) Appl. Environ. Microbiol. 45, 711-713.
- 61 Eisbrenner, G. and Evans, H.J. (1982) J. Bacteriol. 149, 1005-1012.
- 62 Eisbrenner, G. and Evans, H.J. (1982) Plant Physiol. 70, 1667-1672.
- 63 Eisbrenner, G., Hickok, R.E. and Evans, H.J. (1982) Arch. Microbiol. 132, 230-235.
- 64 Ellfolk, N. and Sievers, G. (1965) Acta Chem. Scand. 19, 268-269.
- 65 El Mokadem, M.T. and Keister, D.L. (1982) Israel J. Bot. 31, 102-111.
- 66 Emerich, D.W, Ruiz-Argüeso, T., Ching, T.M. and Evans, H.J. (1979) J. Bacteriol. 137, 153-160.
- 67 Engelke, T., Jagadish, M.N. and Pühler, A. (1987) J. Gen. Microbiol. 133, 3019–3029.
- 68 Evans, H.J., Hanus, F.J., Haugland, R.A., Cantrell, M.A., Xu, L.-S., Russell, S.A., Lambert, G.R. and Harker, A.R. (1983) in World Soybean Research Conference III (Shibles, R., ed.), pp. 935-942, Westview Press, Boulder.
- 69 Evans, H.J., Harker, A.R., Papen, H., Russell, S.A., Hanus, F.J. and Zuber, M. (1987) Annu. Rev. Microbiol. 41, 335-361.
- 70 Falk, J.E., Appleby, C.A. and Porra, R.J. (1959) Symp. Soc. Exp. Biol. 13, 73–86.
- 71 Finan, T.M., Oresnik, I. and Bottacin, A. (1988) J. Bacteriol. 170, 3396-3403.
- 72 Finan, T.M., Wood, J.M. and Jordan, D.C. (1981) J. Bacteriol. 148, 193-202.
- 73 Finan, T.M., Wood, J.M. and Jordan, D.C. (1983) J. Bacteriol. 154, 1403-1413.
- 74 Gardiol, A., Arias, A., Cervenasky, C. and Martinez-Drets, G. (1982) J. Bacteriol. 151, 1621-1623.
- 75 Gennis, R.B. (1986) in Microbial Energy Transduction (Youvan, D.C. and Daldal, F., eds.), pp. 99-103, Cold Spring Harbor Laboratory, Cold Spring Harbor.
- 76 Glenn, A.R. and Brewin, N.J. (1981) J. Gen. Microbiol. 126, 237-241.
- 77 Glenn, A.R. and Dilworth, M.J. (1984) J. Gen. Microbiol. 130, 239-245.
- 78 Glenn, A.R., McKay, J.A., Arwas, R. and Dilworth, M.J. (1984) J. Gen. Microbiol. 130, 239-245.
- 79 Glenn, A.R., Poole, P.J. and Hudman, J.F. (1980) J. Gen. Microbiol. 119, 267-271.
- 80 Gober, J.W. and Kashket, E.R. (1983) J. Bacteriol. 153, 1196-1201.
- 81 Gober, J.W. and Kashket, E.R. (1984) J. Bacteriol. 160, 216-221.
- 82 Gober, J.W. and Kashket, E.R. (1986) J. Bacteriol. 166, 618-622.
- 83 Gober, J.W. and Kashket, E.R. (1987) Proc. Natl. Acad. Sci. USA 84, 4650-4654.
- 84 Godfrey, C.A., Coventry, D.R. and Dilworth, M.J. (1975) in Nitrogen Fixation by Free-Living Microorganisms (Stewart, W.D.P., eds.), pp. 311-332, Cambridge University Press, New York.
- 85 Godfrey, C.A. and Dilworth, M.J. (1971) J. Gen. Microbiol. 69, 385-390.
- 86 Graham, L. and Maier, R.J. (1987) J. Bacteriol. 169, 2555-12560.
- 87 Guerinot, M.L. and Chelm, B.K. (1986) Proc. Natl. Acad. Sci. USA 83, 1837-1841.

- 88 Haaker, H. and Veeger, C. (1977) Eur. J. Biochem. 77, 1-10.
- 89 Hageman, R.V. and Burris, R.H. (1980) Biochim. Biophys. Acta 591, 63-75.
- Harker, A.R., Lambert, G.R., Hanus, F.J. and Evans, H.J. (1984)
 J. Bacteriol. 159, 850-856.
- 91 Harold, F.M. and Papineau, (1972) J. Membr. Biol. 8, 27-44.
- 92 Haugland, R.A., Cantrell, M.A., Beaty, J.S., Hanus, F.J., Russell, S.A. and Evans, H.J. (1984) J. Bacteriol. 159, 1006-1012.
- 93 Hom, S.S.M., Graham, L.A. and Maier, R.J. (1985) J. Bacteriol. 161, 882-887.
- 94 Hom, S.S.M., Novak, P.D. and Maier, R.J. (1988) Appl. Environ. Microbiol. 54, 358-363.
- 95 Hudman, J.F. and Glenn, A.R. (1980) Arch. Microbiol. 128, 72-77.
- 96 Hunt, T.L., Hurst-Calderone, S. and Dayhoff, M.O. (1978) Atlas of Protein Sequence and Structure 5, Suppl. 3, 229-251.
- 97 Huynh, B., Czechowski, M.H., Krüger, H.-J., DerVartanian, D.V., Peck, Jr., H.D. and LeGall, J. (1984) Proc. Natl. Acad. Sci. USA 81, 3728-3732.
- 98 Hyldig-Nielsen, J.J., Jensen, E.Ø., Paludan, K., Wiborg, O., Garrett, R., Jørgensen, P. and Marcker, K. (1982) Nucl. Acids Res. 10.1 689-701.
- 99 Imamura, T., Riggs, A. and Gibson, Q.H. (1972) J. Biol. Chem. 247, 521-526.
- 100 Imperial, J., Ugalde, R.A., Shah, V.K. and Brill, W.J. (1984) J. Bacteriol. 158, 187-194.
- 101 Jacobs, N.J. and Jacobs, J.M. (1975) Biochem. Biophys. Res. Commun. 65, 435-441.
- 102 Jacobs, N.J. and Jacobs, J.M. (1976) Biochim. Biophys. Acta 449,
- 103 Jarrell, K.F. and Sprott, G.D. (1982) J. Bacteriol. 151, 1195-1203.
- 104 Jones, C.W. (1982) Bacterial Respiration and Photosynthesis (Cole, J.A., Knowles, C.J. and Schlessinger, D., eds), American Society for Microbiology, Washington, DC.
- 105 Kagan, S.A. and Brewin, N.J. (1985) J. Gen. Microbiol. 131, 1141-1147.
- 106 Kaltwasser, H. and Frings, W. (1980) in Nickel in the Environment (Nriagu, J.O., ed.), pp. 463-491, John Wiley & Sons, Inc., New York.
- 107 Kashket, E.R. and Barker, S.L. (1977) J. Bacteriol. 130, 1017-1023.
- 108 Keister, D.L. and Marsh, S.S. (1985) in Nitrogen Fixation Research Progress (Evans, H.J., Bottomley, P.J. and Newton, W.E., eds.), p. 348, Martinus Nijhoff, Boston.
- 109 Keister, D.L., Marsh, S.S. and El Mokadem, M.T. (1983) Plant Physiol. 71, 194-196.
- 110 Keithly, J.H. and Nadler, K.D. (1983) J. Bacteriol. 154, 838-845.
- 111 Keyser, H.H., Van Berkum, P. and Weber, D.F. (1982) Plant Physiol. 70, 1626-1630.
- 112 Klemm, D.J. and Barton, L.L. (1985) J. Bacteriol. 164, 316-320.
- 113 Klemm, D.J. and Barton, L.L. (1987) J. Bacteriol. 169, 5209-5215.
- 114 Kranz, R.G. and Haselkorn, R. (1986) Proc. Natl. Acad. Sci. USA 83, 6805-6809.
- 115 Kretovich, V.L., Romanov, V.I. and Korolyov, A.V. (1973) Plant Soil 39, 619-634.
- 116 Kroll, R.G. and Booth, I.R. (1981) Biochem. J. 198, 691-698.
- 117 Kroll, R.G. and Booth, I.R. (1983) Biochem. J. 216, 709-716.
- 118 Kustu, S., Sei, K. and Keener (1986) in Regulation of Gene Expression (Booth, I.R. and Higgins, C.F., eds.), pp. 139-154, Cambridge University Press, Cambridge.
- 119 Kuykendall, L.D. and Elkan, G.H. (1976) Appl. Environ. Microbiol. 32, 511-519.
- 120 Laane, C., Haaker, H. and Veeger, C. (1978) Eur. J. Biochem. 87, 147-153.
- 121 Laane, C., Haaker, H. and Veeger, C. (1979) Eur. J. Biochem. 97, 369-377.
- 122 Laane, C., Krone, W., Konigs, W.N., Haaker, H. and Veeger, C. (1979) FEBS Lett. 103, 328-332.

- 123 Laane, C., Krone, W., Konigs, W. and Veeger, C. (1980) Eur. J. Biochem. 103, 39-46.
- 124 LeGall, J., Ljungdahl, P.O., Moura, I., Peck, Jr., H.D., Xavier, A.V., Moura, J.J.G., Teixeira, M., Huynh, B.H. and Der-Vartanian, D.V. (1982) Biochem. Biophys. Res. Commun. 106, 610-616.
- 125 Lehtovaara, P. and Ellfolk, N. (1975) Acta Chem. Scand. 29 (Ser. B), 56-60.
- 126 Leong, S.A., Ditta, G.S. and Helinski, D.R. (1982) J. Biol. Chem. 257, 8724–8730.
- 127 Leong, S.A., Williams, P.H. and Ditta, G.S. (1985) Nucl. Acids Res. 13, 5965-5976.
- Leyva, A., Palacios, J.M., Mozo, T. and Ruiz-Argueso, T. (1987)
 J. Bacteriol. 169, 4929–4934.
- 129 Lim, S.T. (1985) Arch. Microbiol. 142, 393-396.
- 130 Magasanik, B. (1982) Annu. Rev. Genet. 16, 135-168.
- 131 Maier, R.J. (1981) J. Bacteriol. 145, 533-540.
- 132 Maier, R.J., Campbell, N.E.R., Hanus, F.J., Simpson, F.B., Russell, S.A. and Evans, H.J. (1978) Proc. Natl. Acad. Sci. USA 75, 3258-3262.
- 133 Maier, R.J., Graham, L., Keefe, R.G., Pihl, T. and Smith, E. (1987) J. Bacteriol. 169, 2548-2554.
- 134 Maier, R.J., Hanus, F.J. and Evans, H.J. (1979) J. Bacteriol. 137, 824–829.
- 135 Maier, R.J. and Merberg, D.M. (1982) J. Bacteriol. 150, 161-167.
- 136 Matus, V.K., Melik-Sarkisyan, S.S. and Kretovitch, V.L. (1973) Microbiology (U.S.S.R.) 42, 95-100.
- 137 Mayer, S.M., Beale, S.I. and Weinstein, J.D. (1987) J. Biol. Chem. 262, 12541-12549.
- 138 McAllister, C.F. and Lepo, J.E. (1983) J. Bacteriol. 153, 1155-1162.
- 139 McClung, C.R., Somerville, J.E., Guerinot, M.L. and Chelm, B.K. (1987) Gene 54, 133-139.
- 140 Mckay, I.A., Glenn, A.R. and Dilworth, M.J. (1985) J. Gen. Microbiol. 131, 2067-2073.
- 141 Merberg, D. and Maier, R.J. (1983) Science 220, 1064-1065.
- 142 Merberg, D. and Maier, R.J. (1984) J. Bacteriol. 160, 448-450.
- 143 Merberg, D., O'Hara, E.B. and Maier, R.J. (1983) J. Bacteriol. 156, 1236-1242.
- 144 Minchin, F.R., Sheehy, J.E. and Witty, J.F. (1985) in Nitrogen Fixation Research Progress (Evans, H.J., Bottomley, P.J. and Newton, W.E., eds.), pp. 285-291, Martinus Nijhoff, Boston.
- 145 Mori, M. and Sano, S. (1968) Biochem. Biophys. Res. Commun. 32, 610-615.
- 146 Mortenson, L.E. and Thornley, R.N.F. (1979) Annu. Rev. Biochem. 48, 387-418.
- 147 Moshiri, F. and Maier, R.J. (1988) J. Biol. Chem. 263, 17809–17816.
- 148 Nadler, K.D. (1981) in Current Perspectives in Nitrogen Fixation (Gibson, A.H. and Newton, W.E., eds.), p. 414. Australian Academy of Science, Canberra.
- 149 Nadler, K.D. and Avissar, Y.J. (1977) Plant Physiol. 60, 433-436.
- 150 Nakamura, T., Tokuda, H. and Unemoto, T. (1984) Biochim. Biophys. Acta 776, 330-336.
- 151 Nautiyal, C.S., Hegde, S.V. and Van Berkum, P. (1988) Appl. Environ. Microbiol. 54, 94-97.
- 152 Nautiyal, C.S., Van Berkum, P. and Keister, D.L. (1987) 11th North American Rhizobium Conference, Laval, Abstr. PP4.
- 153 Nelson, L.M. and Salminen, S.O. (1982) J. Bacteriol. 151, 989–995.
- 154 Ninfa, A.J. and Magasanik, B. (1986) Proc. Natl. Acad. Sci. USA 83, 5909-5913.
- 155 Nixon, B.T., Ronson, C.W. and Ausubel, F.M. (1986) Proc. Natl. Acad. Sci. U.S.A. 83, 7850-7854.
- 156 Novak, P.D. (1988) Ph.D. Thesis, The Johns Hopkins University.
- 157 Novak, P.D. and Maier, R.J. (1987) J. Bacteriol. 169, 2708-2712.
- 158 O'Brian, M.R., Kirshbom, P.M. and Maier, R.J. (1987) J. Bacteriol. 169, 1089-1094.

- 159 O'Brian, M.R., Kirshbom, P.M. and Maier, R.J. (1987) Proc. Natl. Acad. Sci. USA 84, 8390-8393.
- 160 O'Brian, M.R. and Maier, R.J. (1982) J. Bacteriol. 152, 422-430.
- 161 O'Brian, M.R. and Maier, R.J. (1983) J. Bacteriol. 155, 481-487.
- 162 O'Brian, M.R. and Maier, R.J. (1985) J. Bacteriol. 161, 507-514.
- 163 O'Brian, M.R. and Maier, R.J. (1985) J. Bacteriol. 161, 775-777.
- 164 O'Brian, M.R. and Maier, R.J. (1987) Proc. Natl. Acad. Sci. USA 84, 3219-3223.
- 165 O'Brian, M.R. and Maier, R.J. (1988) Adv. Microbiol. Physiol. 29, 1-52.
- 166 Orme-Johnson, W.E. (1985) Annu. Rev. Biophys. Biophys. Chem. 14, 419–459.
- 167 Parke, D. and Ornston, L.N. (1984) J. Gen. Microbiol. 130, 1743-1750.
- 168 Partridge, C.D.P. and Yates, M.G. (1982) Biochem. J. 204, 339-344.
- 169 Patterson, T.G., Peterson, J.B. and LaRue, T.A. (1983) Plant Physiol. 70, 695-700.
- 170 Pawlowski, K., Schell, J. and De Bruijn, F.J. (1988) 7th International Congress on Nitrogen Fixation, 13-20 March, Cologne, Abstr 9-37
- 171 Peterson, J.B. and LaRue, T.A. (1981) Plant Physiol. 68, 489-493.
- 172 Pienkos, P.T. and Brill, W.J. (1981) J. Bacteriol. 145, 743-751.
- 173 Porra, R.J. (1975) Anal. Biochem. 68, 289-298.
- 174 Postgate, J.R. (1982) The Fundamentals of Nitrogen Fixation, Cambridge University Press, Cambridge.
- 175 Rawsthorne, S. and LaRue, T.A. (1986) Plant Physiol. 81, 1097-1102.
- 176 Reibach, P.H. and Streeter, J.G. (1984) J. Bacteriol. 159, 47-52.
- 177 Rivera-Ortiz, J.M. and Burris, R.H. (1975) J. Bacteriol. 123, 537-545.
- 178 Roessler, P.G. and Nadler, K.D. (1982) J. Bacteriol. 149, 1021–1026.
- 179 Romanov, V.I., Matus, V.K., Korolov, A.V. and Kretovitch, V.L. (1973) Microbiology (U.S.S.R.) 42, 871-876.
- 180 Ronson, C.W. and Astwood, P.M. (1985) in Nitrogen Fixation Research Progress (Evans, H.J., Bottomley, P.J. and Newton, W.E., eds.), pp. 201-207, Martinus Nijhoff, Boston.
- 181 Ronson, C.W., Astwood, P.M. and Downie, J.A. (1984) J. Bacteriol. 160, 903-909.
- 182 Ronson, C.W., Astwood, P.M., Nixon, B.T. and Ausubel, F.M. (1987) Nucl. Acids Res. 15, 7921-7934.
- 183 Ronson, C.W., Lyttleton, P. and Robertson, J.G. (1981) Proc. Natl. Acad. Sci. USA 78, 4284–4288.
- 184 Ronson, C.W., Nixon, B.T., Albright, L.M. and Ausubel, F.M. (1987) J. Bacteriol. 169, 2424-2431.
- 185 Ronson, C.W., Nixon, B.T. and Ausubel, F.M. (1987) Cell 49, 579–581.
- 186 Ronson, C.W. and Primrose, S.B. (1979) J. Gen. Microbiol. 112, 77-88.
- 187 Salminen, S.O. and Streeter, J.G. (1987) J. Bacteriol. 169, 495-499.
- 188 Salminen, S.O. and Streeter, J.G. (1987) Plant Physiol. 83, 535-540.
- 189 San Francisco, M.J.D. and Jacobson, G.R. (1985) J. Gen. Microbiol. 131, 765-773.
- 190 Sapshead, L.M. and Wimpenny, J.W.T. (1972) Biochim. Biophys. Acta 267, 388–397.
- 191 Saroso, S., Dilworth, M.J. and Glenn, A.R. (1986) J. Gen. Microbiol. 132, 243-249.
- 192 Saroso, S., Glenn, A.R. and Dilworth, M.J. (1984) J. Gen. Microbiol. 130, 1809–1814.
- 193 Sasaki, T., Motokawa, Y. and Kikuchi, G. (1970) Biochim. Biophys. Acta 197, 284-291.
- 194 Schubert, K.R., Engelke, J.A., Russell, S.A. and Evans, H.J. (1977) Plant Physiol. 60, 651-654.
- 195 Schubert, K.R. and Evans, H.J. (1976) Proc. Natl. Acad. Sci. USA 73, 1207–1211.

- 196 Seefeldt, L.C., Fox, C.A. and Arp, D.J. (1986) J. Biol. Chem. 261, 10688–10694.
- 197 Shah, V.K., Uglade, R.A., Imperial, J. and Brill, W.J. (1984) Annu. Rev. Biochem. 53, 231-257.
- 198 Sidloi-Lumbroso, R., Kleiman, L. and Schulman, H.M. (1978) Nature (London) 273, 558-560.
- 199 Simpson, F.B. and Burris, R.H. (1984) Science 224, 1095-1097.
- 200 Skulachev, V.P. (1978) FEBS Lett. 87, 171-179.
- 201 Sone, N. and Yanagita, Y. (1982) Biochim. Biophys. Acta 682, 216–226.
- 202 Spiro, S. and Guest, J.R. (1987) J. Gen. Microbiol. 133, 3279–3288.
- 203 Stam, H., van Verseveld, H.W., De Vries, W. and Stouthamer, A. (1984) Arch. Microbiol. 139, 53-60.
- 204 Stowers, M.D. (1985) Annu. Rev. Microbiol. 39, 89-108.
- 205 Stults, L.W., Mallick, S. and Maier, R.J. (1987) J. Bacteriol. 169, 1398–1402.
- 206 Stults, L.W., Moshiri. F.M. and Maier, R.J. (1986) J. Bacteriol. 166, 795–800.
- 207 Stults, L.W., O'Hara, E.B. and Maier, R.J. (1984) J. Bacteriol. 159, 153-158.
- 208 Stults, L.W. Sray, W.A. and Maier, R.J. (1986) Arch. Microbiol. 146, 280–283.
- 209 Tedeschi, H. (1980) Biol. Rev. 55, 171-206.
- 210 Thauer, R.K., Diekert, G. and Schonheit, P. (1980) Trends Biochem. Sci. 5, 304–306.
- 211 Thummler, F. and Verma, D.P.S. (1987) J. Biol. Chem. 262, 14730–14736.
- 212 Tjepkema, J.D. (1983) Can. J. Bot. 61, 2924-2929.
- 213 Tjepkema, J.D. and Yocum, C.S. (1974) Planta 119, 351-360.
- 214 Tokuda, H., Nauamura, T. and Unemoto, T. (1981) Biochemistry 20, 4198–4203.
- 215 Trinchant, J.C., Birot, A.M., Denis, M. and Rigaud, J. (1983) Arch. Microbiol. 134, 182–186.
- 216 Trinchant, J.C., Birot, A.M. and Rigaud, J. (1981) J. Gen. Microbiol. 125, 159–165.
- 217 Trinchant, J.-C. and Rigaud, J. (1979) Physiol. Vég. 17, 547-556.
- 218 Truelsen, T.A. and Wyndaele, R. (1984) Physiol. Planta. 62, 45–50.
- 219 Tuzimura, K. and Meguro, H. (1960) J. Biochem. 47, 391-397.
- 220 Tuzimura, K. and Watanabe, I. (1964) Plant Cell Physiol. 5, 157-170.
- 221 Uheda, E. and Syono, K. (1982) Plant Cell Physiol. 23, 85-90.
- 222 Van Berkum, P. (1987) J. Bacteriol. 169, 4565-4569.
- 223 Van Berkum, P. (1988) Annu. Meeting Am. Soc. Microbiol., Miami Beach, Abstr. MK71.
- 224 Van Berkum, P. and Maier, R.J. (1988) J. Bacteriol. 170, 1962–1964.
- 225 Verma, D.P.S., Nash, D.T. and Schulman, H.M. (1974) Nature (London) 251, 74-77.
- 226 Watson, R.J., Yiu-Kwok, C., Wheatcroft, R., Yang, A.-F. and Han, S. (1988) J. Bacteriol. 170, 927-934.
- 227 Williams, R.J.P. (1978) Biochim. Biophys. Acta 505, 1-44.
- 228 Williams, R.J.P. (1978) FEBS Lett. 85, 9-19.
- 229 Winans, S.C., Ebert, P.R., Stachel, S.E., Gordon, M.P. and Nester, E.W. (1986) Proc. Natl. Acad. Sci. USA 83, 8278-8282.
- 230 Wittenberg, J.B. (1970) Physiol. Rev. 50, 559-636.
- 231 Wittenberg, J.B., Appleby, C.A. and Wittenberg, B.A. (1972) J. Biol. Chem. 247, 527–531.
- 232 Wittenberg, J.B., Bergensen, F.J., Appleby, C.A. and Turner, G.L. (1974) J. Biol. Chem. 249, 4057–4066.
- 233 Yamamoto, N. and Droffner, M.L. (1985) Proc. Natl. Acad. Sci. USA 2077-2081.
- 234 Zuber, M., Harker, A.R., Sultana, M.A. and Evans, H.J. (1986) Proc. Natl. Acad. Sci. USA 83, 7668-7672.
- 235 Martinez-de Drets, G. and Arias, A. (1972) J. Bacteriol. 109, 467-470.

- 236 Jensen, E.Ø., Marcker, K.A. and Villadsen, I.S. (1986) EMBO J. 5, 843-847.
- 237 Jensen, E.Ø., Marcker, K.A., Schell, J. and De Bruijn, F.J. (1988) EMBO J. 7, 1265-1271.
- 238 Bogusz, D., Appleby, C.A., Landsmann, J., Dennis, E.S., Trinick, M.J. and Peacock, W.J. (1988) Nature 331, 178-180.
- 239 Stanley, J., Dowling, D.N. and Broughton, W.J. (1988) Mol. Gen. Genet. 215, 32-37.
- 240 Jacobs, N.J., Borotz, S.E. and Guerinot, M.L. (1989) J. Bacteriol. 171, 573-576.
- 241 Sayavedra-Soto, L.A., Powell, G.K., Evans, H.J. and Morris, R.O. (1988) Proc. Natl. Acad. Sci. USA 85, 8395-8399.
- 242 Maier, R.J. and Graham, L. (1988) J. Bacterial. 170, 5613-5619.
- 243 Soberon, M., Williams, H.D., Poole, R.K. and Escamilla, E. (1989) J. Bacteriol. 171, 465-472.
- 244 Bothe, H., De Bruijn, F.J. and Newton, W.E. (1988) Nitrogen Fixation: Hundred Years After, 878 pp., Gustav Fischer, New York.